

## Introduction

Gas chromatography (GC) is a separation technique for volatile compounds that involves the vaporization of a sample through an injector. Then, the sample is carried by a gas moving through a stationary phase (column) packed with a gel or a solid until reaching a detector. The CME Department recently acquired a GC system with exchangeable detectors: Thermal conductivity detector (TCD) and flame ionization detector (FID). *To enhance interdisciplinary collaboration across Units and Departments within the University of Dayton, this work seeks to develop GC and Headspace-GC (HS-GC) analytical methods for use both in the classroom and in research.*

## Objectives

- To understand the variables manipulated in GC and HS-GC and apply this knowledge to developing characterization methods
- To generate a characterization method to determine the concentration of *ethanol in aqueous mixtures obtained from a distillation process*

## Methods

An extensive literature review has been conducted through Journal articles, books, and webinars before starting with the Summer Undergraduate Research Experience(SURE) program with the SoE this summer.

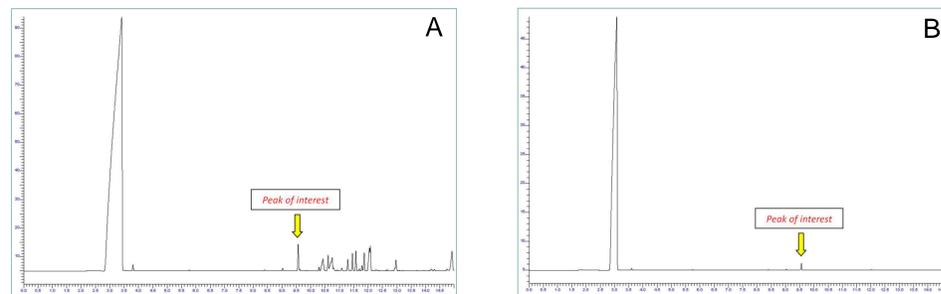
## Equipment

- Thermo Scientific GC w/TCD and FID detector
- Autosampler (Triplus RSH)
- Capillary Column [TG-BOND Q+]
- HS and GC vials



## Background: GC vs. HS-GC

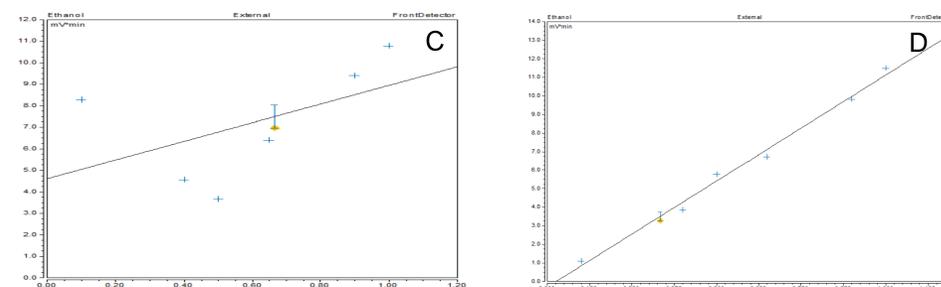
Headspace sampling: The more volatile compounds are extracted from the sample. For example:



Perfume Chromatographs obtained using a direction injection GC (A) method and a Headspace sampling technique (B). (Tipler, 2013)

## Results and Discussion

- Two separate calibration curves (C,D) were generated in the laboratory using liquid samples of 10, 40, 50, 65, 90, and 100 % (v/v) ethanol/water using GC.
- Sample preparation was improved between analysis as shown on each calibration



EtOH/H<sub>2</sub>O Calibration Curves

## Proposed HS-GC-Method

A method proposed by Shan et al., (2011) using a HS-GC with a FID detector will be tested during the summer to evaluate SCFA.

## Headspace Explanation

- Partition coefficient:  $K=C_s/C_g$
- Phase Ratio:  $B=V_g/V_s$
- Headspace Concentration:  $C_g=C_o/(K+B)$
- As T goes up, K goes down
- Activity coefficient:  $P_i=P_o \times Y_i \times X_s(i)$
- Adding salts to EtOH/H<sub>2</sub>O samples

## Using HS-GC Quantitatively

- Sample volume:  $V_s=V_e \times (F_{col}/F_{col}+F_s)$
- Injection time:  $T_i=(V_{syr}/F_{col}+F_s) \times (P_i/P_a)$
- This combination allows sample concentrations determination



Thermo Scientific GC  
And Autosampler



Distillation Column

## Future Applications

- Analyze ethanol and water samples obtained from the Unit Operations Laboratory.
- Characterize the efficiency of GC analysis of EtOH/H<sub>2</sub>O/oil samples vs. using HS-GC for ternary mixtures
- Characterize SCFA (short chain fatty acid) content in *Listeria* metabolites present in mice feces with the SURE program in the SoE during this summer

## Acknowledgements

We are grateful to the Chemical and Materials Engineering Department for supporting this work and the 2018 SURE SoE program.

## Selected References

- Kolb, B., & Ettre, L. S. (2006). *Static headspace-gas chromatography: theory and practice*. John Wiley & Sons.
- Shan, X., Tiscione, N. B., Alford, I., & Yeatman, D. T. (2011). A study of blood alcohol stability in forensic antemortem blood samples. *Forensic science international*, 211(1-3), 47-50.
- Tipler, Andrew. "An Introduction to Headspace Sampling in Gas Chromatography Fundamentals and Theory." (2013)
- Tiscione, N. B., Alford, I., Yeatman, D. T., & Shan, X. (2011). Ethanol analysis by headspace gas chromatography with simultaneous flame-ionization and mass spectrometry detection. *Journal of analytical toxicology*, 35(7), 501-511..