

Rationale

- Silver nanoparticles (AgNPs) exhibit unique properties due to their size and are becoming increasingly prevalent in consumer and industrial settings¹
- AgNPs are known to induce cytotoxicity within in vitro and in vivo environments²
- Correlations have been found between bioresponses and concentration of internalized NPs³
- The goal was to develop pharmacokinetic profiles for experimental AgNPs within a lung (A549) cell model

Methodology

- **Cell Model:** Human alveolar cells (A549)
- **10 nm AgNPs:** citrate and PVP coatings
- **Cytotoxicity:** Exposed A549s to AgNPs and analyzed viability using MTS assay
- **Pharmacokinetics:** Exposure of A549s to AgNPs and quantified delivered dose via absorbance analysis

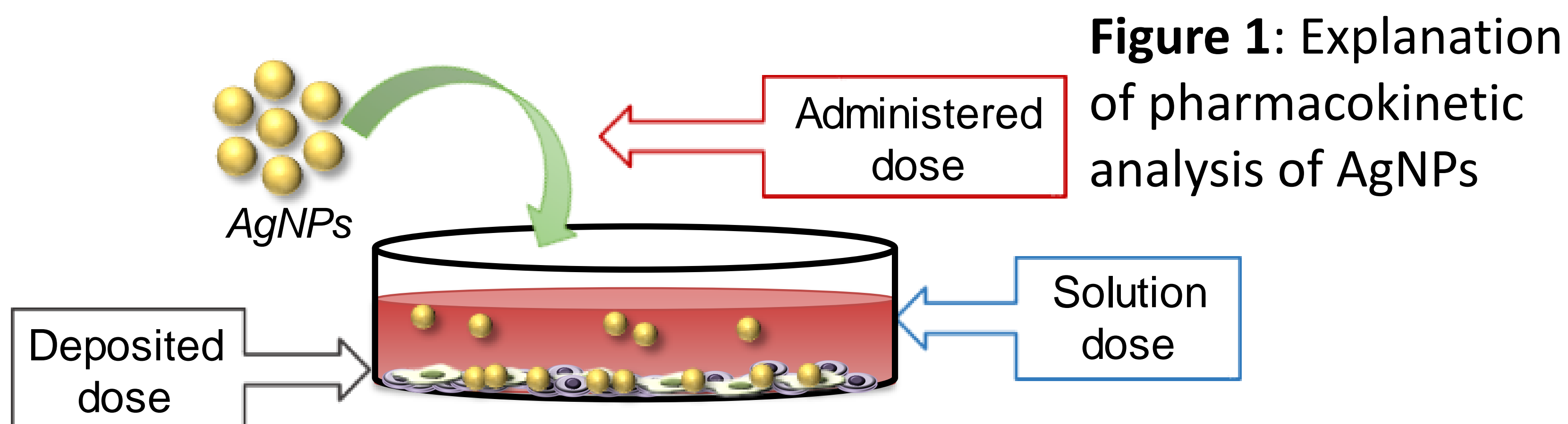


Figure 1: Explanation of pharmacokinetic analysis of AgNPs

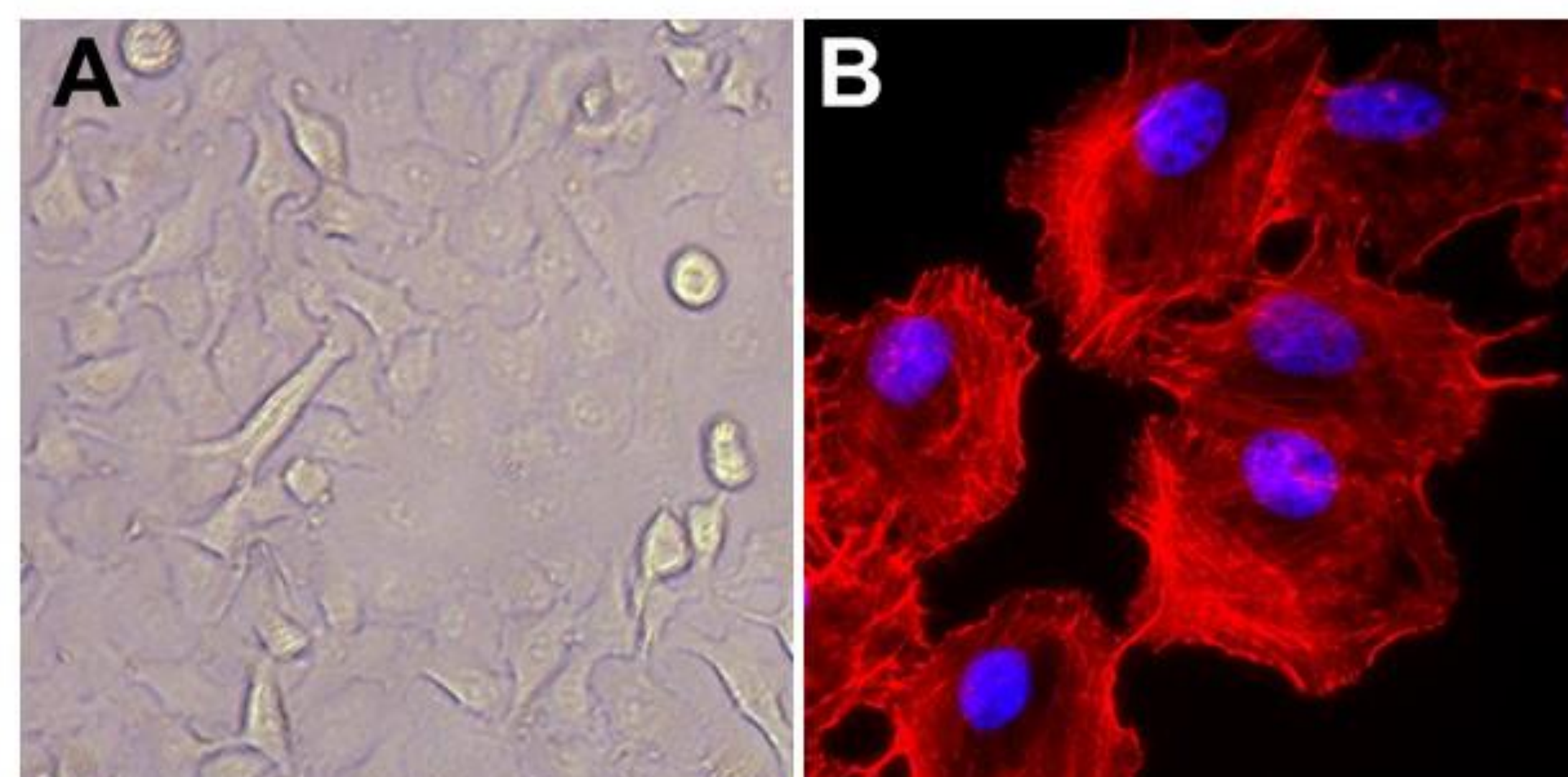


Figure 2: Images of A549 cells via (A) light and (B) fluorescence microscopy

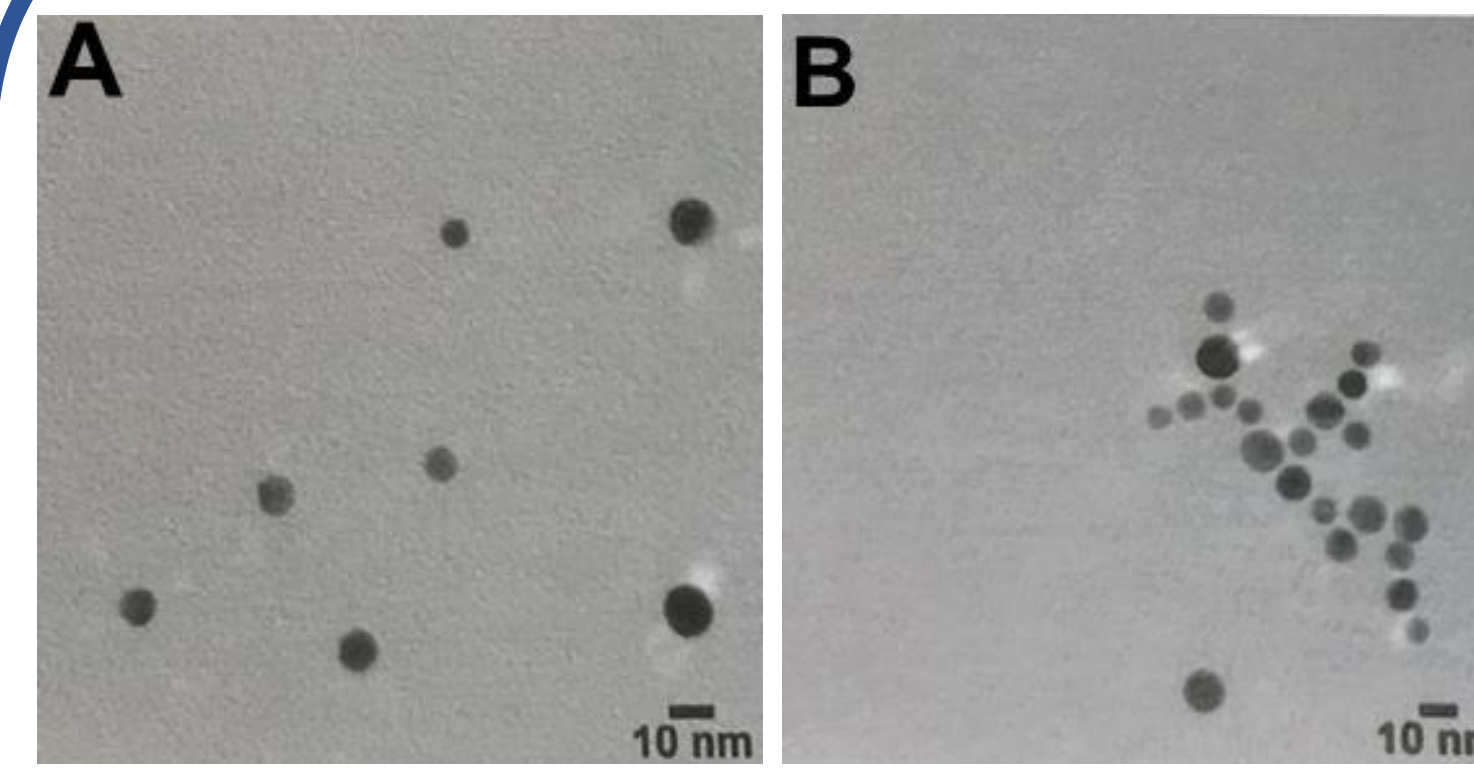


Figure 3: TEM images of (A) citrate and (B) PVP AgNPs

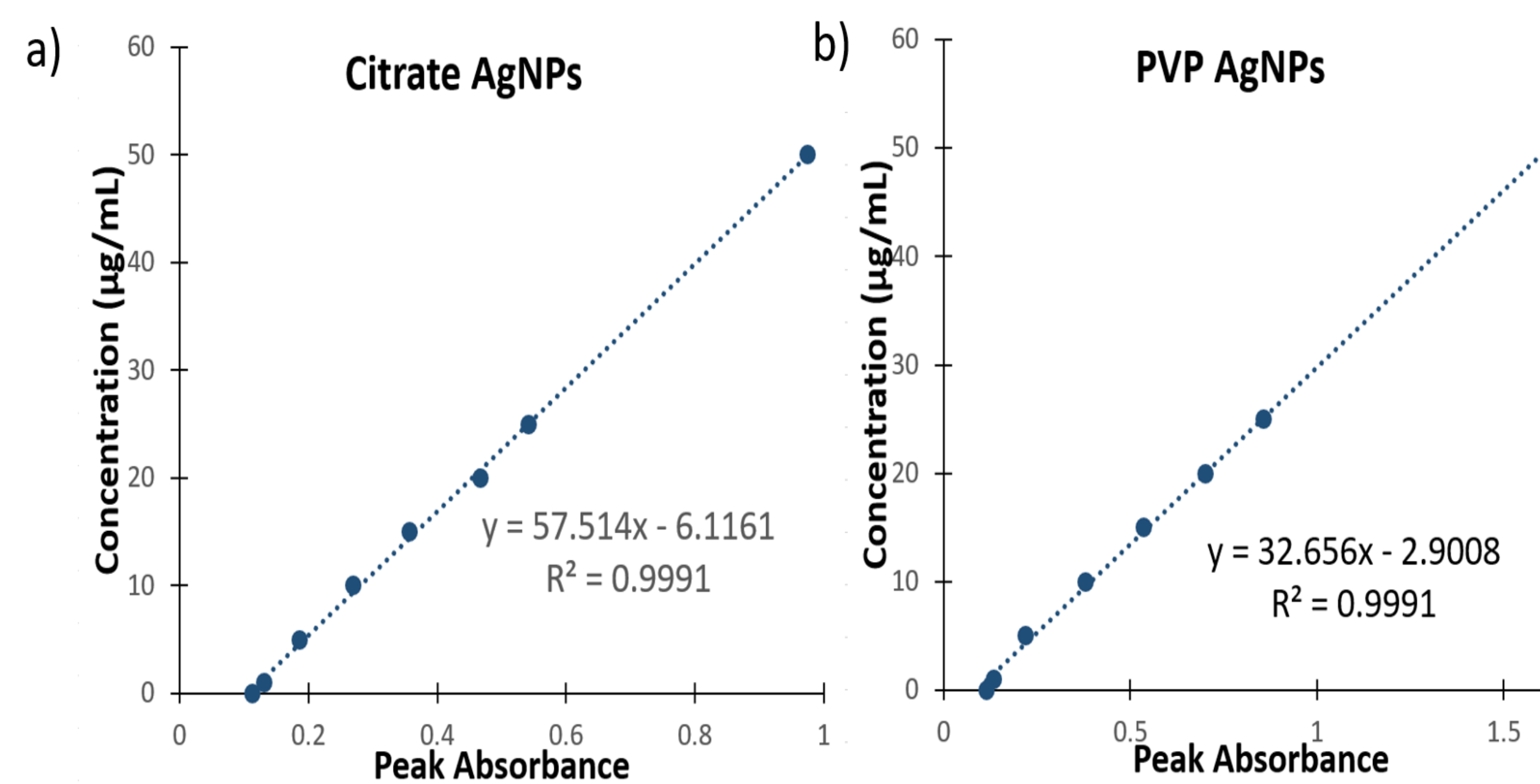


Figure 5: Calibration curves correlated AgNP dose to peak absorbance

AgNPs interact with the A549 cells to a high degree within a static environment.

Results

Table 1: AgNP Characterization

	Primary Size (nm)	Agglomerate Size (nm)		Zeta Potential (mV)	
		Water	Media	Water	Media
Citrate	9.3 ± 1.8	22.9 ± 2.3	47.0 ± 2.9	-38.6 ± 1.4	-9.4 ± 0.7
PVP	10.6 ± 1.6	19.1 ± 1.5	35.3 ± 1.8	-22.1 ± 1.1	-10.2 ± 0.6

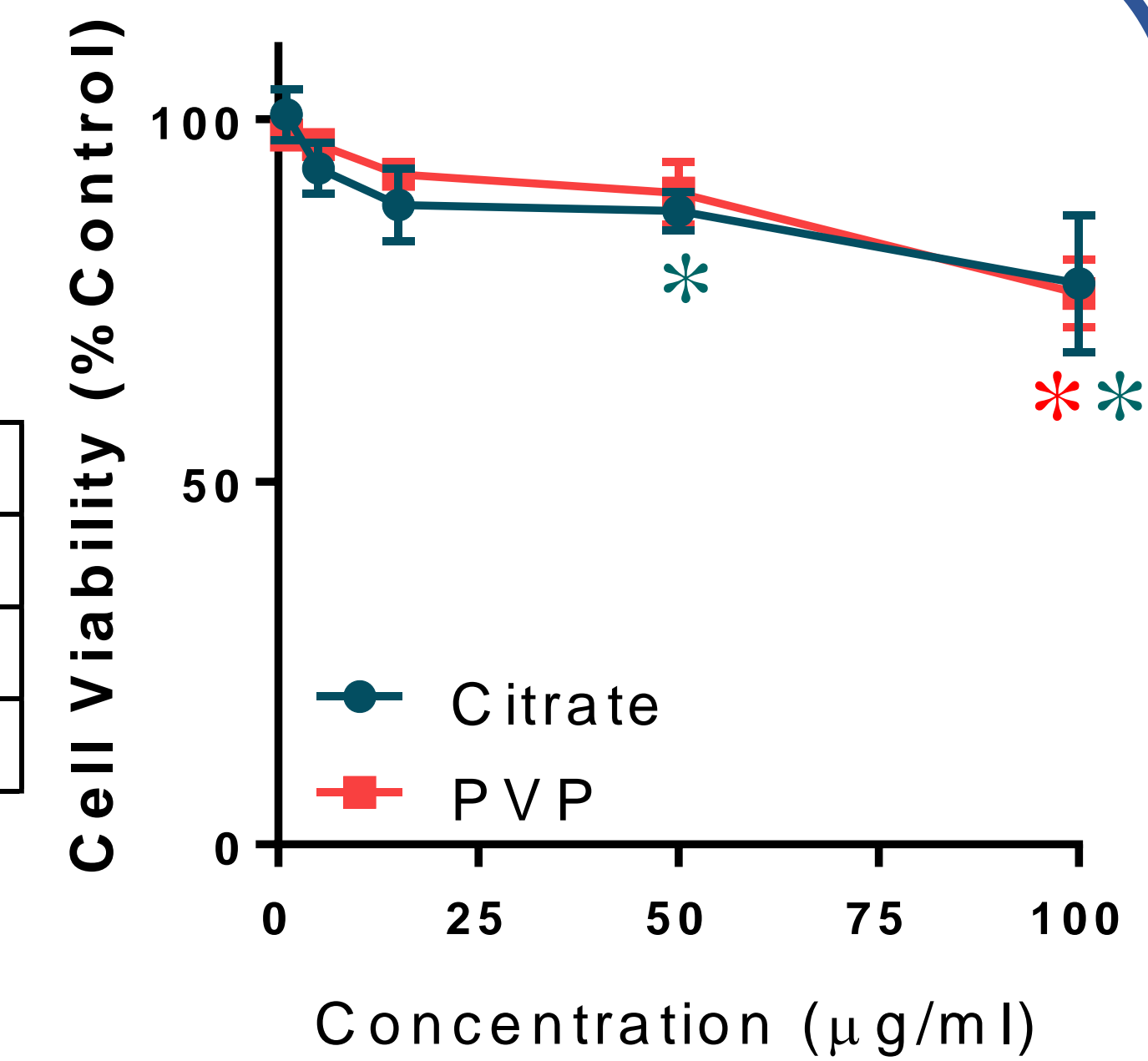


Figure 4: Toxicity analysis determined AgNP dosage for deposition experiments.

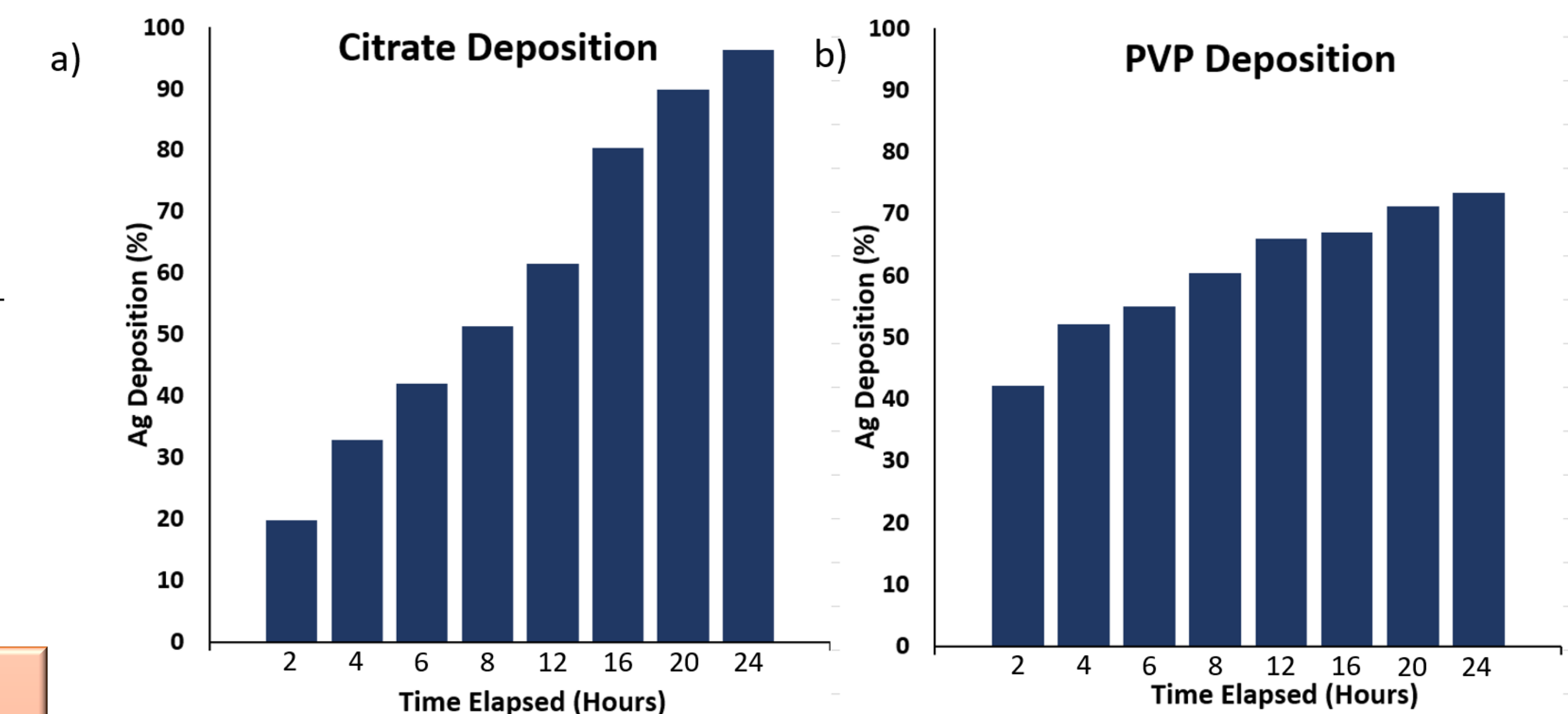


Figure 6: AgNP deposition over a range of time exposures from 2-24 hours

Conclusions

- Using the toxicity analysis, an AgNP dosage of 25 µg/mL was selected.
- A high degree of A549-AgNP association was determined for both experimental particle sets, with citrate coated AgNPs having a higher degree of interaction than PVP coated AgNPs.
- Future work will repeat these analyses at different nanoparticle sizes and under dynamic exposure conditions.

Acknowledgements and References

Portions of this work were funded through the University of Dayton Summer Undergraduate Research Experience Program, the National Science Foundation, Colgate Palmolive, and the Verffhoff family.

1. V. Bastos, et al. *J Nanopart Res*, **2017**, 19, 163.
2. K.K. Comfort, et al. *ACS Nano*, **2014**, 8, 3260.
3. J.G. Teeguareden et al. *Toxicol. Sci.*, **2007**, 95, 300.